

**D.M.PUBLIC SCHOOL** 

## PARSIA, PUTKI, DHANBAD

# Fortnightly Syllabus Planning (2024-25)

## Class: X

#### Subject: CHEMISTRY

	New Session begins on 4 <sup>th</sup> April, 2024						
SI No	Duration	No. of Teaching Days	Syllabus to be covered				
1	5 <sup>th</sup> April - 15 <sup>th</sup> April	7	CHAPTER- 1 Chemical Reactions and Equations Introduction Defination, Examples				
2	16 <sup>th</sup> April – 30 <sup>th</sup> April	7	The Characterstics of Chemical reactions with Examples, balancing of chemical Equations,				
3	1 <sup>st</sup> May – 15 <sup>th</sup> May	7	Types of Chemical Reaction with Examples and Reactions,				
4	16 <sup>th</sup> June – 30 <sup>st</sup> June	7	Redox reactions in everyday Life, Rancidity, Prevention from rancidity (Revision )				
5	1 <sup>st</sup> July – 9 <sup>th</sup> July - PRE MID TERM EXAMINATION						
6	10 <sup>th</sup> July – 31 <sup>th</sup> July	14	CHAPTER- 2 Acid base and Salts, Introduction, Indicators, types of Indicators,				
7	1 <sup>st</sup> Aug – 15 <sup>th</sup> Aug	7	Types of acids, Their Origin, Differnce between Organic and Inorganic acids,				
8	16 <sup>th</sup> Aug – 31 <sup>st</sup> Aug	7	Bases- Types of Bases, Difference between acids and bases, Concepts of pH scale,				
9	1 <sup>st</sup> Sep – 12 <sup>th</sup> Sep	7	preparation and uses of Sodium Hydroxide, Bleaching powder, Baking soda, Washing soda and Plaster of Paris. <b>Revision</b>				
10	13 <sup>st</sup> Sep- 1 <sup>th</sup> Oct - MID 1						
11	2 <sup>st</sup> Oct – 15 <sup>th</sup> Oct	7	CHAPTER-3 Metals and Non Metals, Introduction, Physical properties of metals and Non Metals,				
12	16 <sup>th</sup> Oct – 31 <sup>st</sup> Oct	7	The Chemical properties of metals and Non Metaks , Concepts of Metalloids,Reaction of Metals with acids and waters,				
13	1 <sup>st</sup> Nov – 15 <sup>th</sup> Nov	7	Reactivity Series, Properties of Ionic Compounds, Extractions of Metals				

14	16 <sup>th</sup> Nov – 30 <sup>th</sup> Nov	7	The concept of Corrosion, Prevention from corrosion, Revision.
15	2 <sup>st</sup> Dec- 10 <sup>th</sup> Dec - POST	<b>MID TERM</b>	EXAMINATION
16	11 <sup>th</sup> Dec – 24 <sup>st</sup> Dec	7	CHAPTER-04 Carbon and Its Compounds, Introduction, The Covalant Compounds, Types of covalent compounds
17	3 <sup>st</sup> Jan — 15 <sup>th</sup> Jan	7	<b>Carbon and Its Allotropes, versatile nature of carbon,</b> Nomenclature of carbon compounds containing functional groups (halogens, alcohol, ketones, aldehydes, alkanes and alkynes), difference between saturated hydro carbons and unsaturated hydrocarbons.
18	16 <sup>th</sup> Jan – 31 <sup>st</sup> Jan	7	Chemical properties of carbon compounds (combustion, oxidation, addition and substitution reaction). Ethanol and Ethanoic acid (only properties and uses), soaps and detergents.
19	1 <sup>st</sup> Feb – 15 <sup>th</sup> Feb	7	Revision NCERT Intext and excercise questions
20		2	0 <sup>th</sup> Feb to 8 <sup>th</sup> March: Annual Examination

#### **Portion For Assessments**

Assessment	Portion
PRE MID TERM (PT 1)	Chemical reactions and equations
MID TERM	Chemical reactions and equations
	Acid base and Salts
POST MID TERM (PT 2)	Metals and Non Metals
ANNUAL	Chemical reactions and equations Acid base and Salts Metals and Non Metals Carbon and its compounds